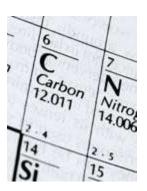
Introduction to the Periodic Table

Atomic Number • Symbol • Atomic Weight

Element • Compound • Mixture

What is the PERIODIC TABLE?

- o Shows all known elements in the universe
- o Organizes the elements by chemical properties.



How do you read the PERIODIC TABLE?

6 Carbon 12,01

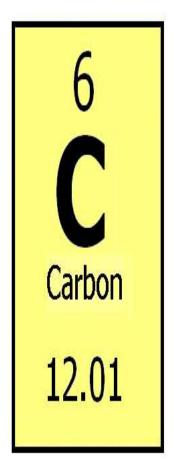
___ Atomic number

— Symbol

— Name

___ Atomic Weight

What is the ATOMIC NUMBER?



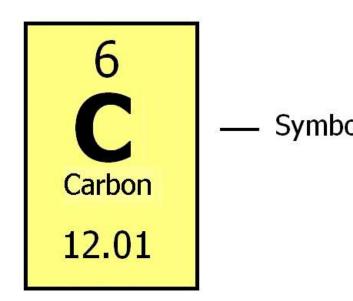
Atomic number

- o The number of protons found in the nucleus of an atom

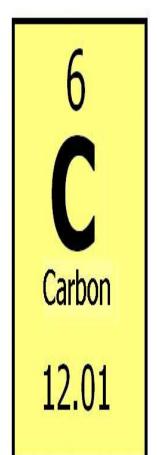
 Or
- o The number of electrons surrounding the nucleus of an atom.

What is the SYMBOL?

o An abbreviation of the element name.



What is the ATOMIC WEIGHT?



o The number of protons and neutrons in the nucleus of an atom.

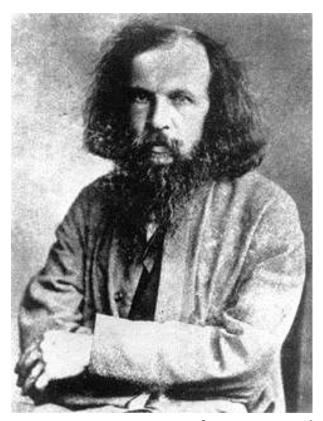
Atomic Weight

How do I find the number of protons, electrons, and neutrons in an element using the periodic table?

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o # of PROTONS = ATOMIC NUMBER
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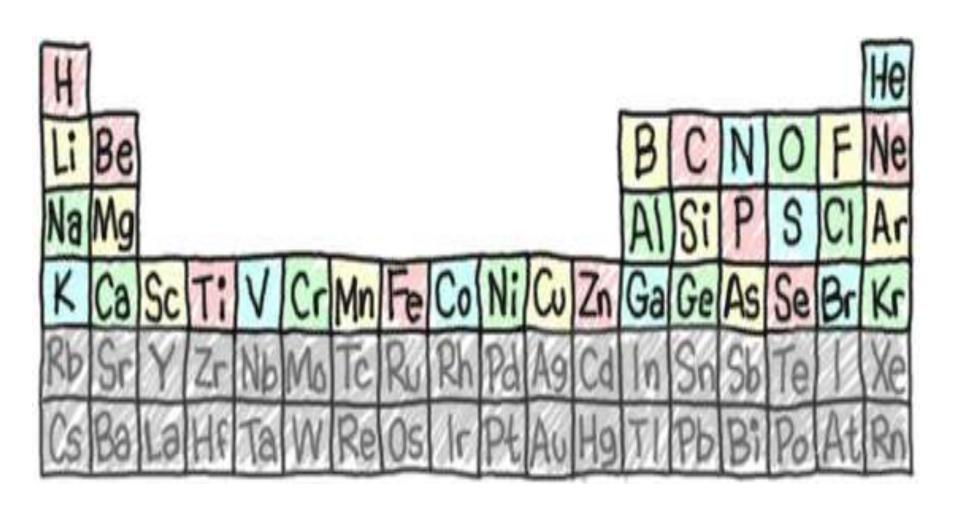
o # of ELECTRONS = ATOMIC NUMBER

Now you are almost as smart as I am! I am Dmitri Mendeleev!



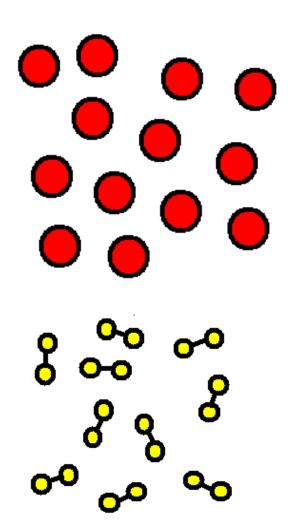
But not as handsome! Man, I look GOOD!

Elements, Compounds, and Mixtures

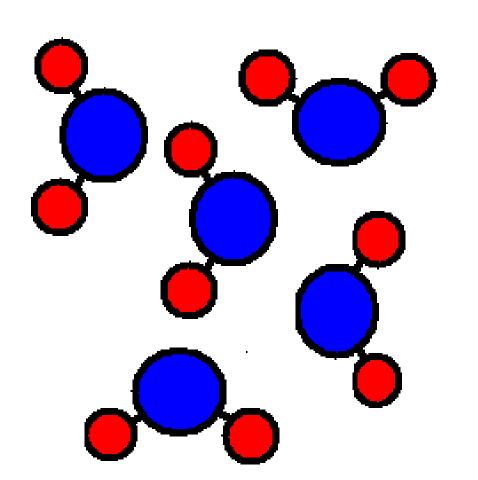


What is an ELEMENT?

- o A substance composed of a single kind of atom.
- o Cannot be broken down into another substance by chemical or physical means.



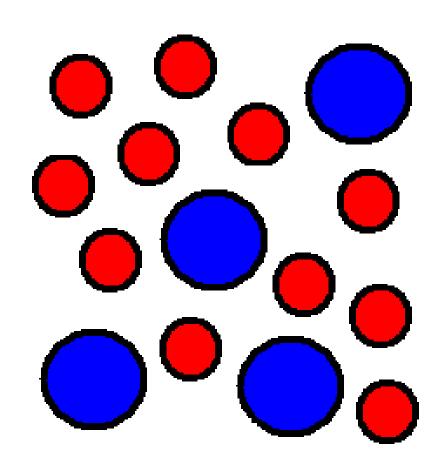
What is a COMPOUND?



o A substance in which two or more different elements are CHEMICALLY bonded together.

What is a MIXTURE?

o Two or more substances that are mixed together but are NOT chemically bonded.



Element, Compound or Mixture?

